

Buckled Two-Dimensional Xene Sheets

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Abstract

35 Silicene, germanene, and stanene, are part of a monoelemental class of two-dimensional
36 (2D) crystals termed 2D-Xenes (X=Si, Ge, Sn) which, together with their ligand-
37 functionalized derivatives referred to as Xanes, are comprised of group IVA atoms
38 arranged in a honeycomb lattice – similar to graphene but with varying degrees of
39 buckling. Their electronic structure range from trivial insulators, to semiconductors with
40 tunable gaps, to semimetallic, depending on the substrate, chemical functionalization, and
41 strain. More than a dozen different topological insulator states are predicted to emerge,
42 including the quantum spin Hall state at room temperature, which, if realized, would
43 enable new classes of nanoelectronic and spintronic devices, such as the topological field
44 effect transistor. The electronic structure can be tuned, for example, by changing the
45 group IVA element, the degree of spin-orbit coupling, the functionalization chemistry, or
46 the substrate, making the 2D-Xene systems promising multifunctional 2D materials for
47 nanotechnology. This Perspective Article highlights the current state-of-the art and future
48 opportunities in the manipulation and stability of these materials, their functions and
49 applications, and novel device concepts.

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The rise of graphene has inspired the flourish of a so-called “flatland” of two-dimensional (2D) materials with complementary properties and functionalities enabling rapid advances in science and engineering, and an accelerated global development in nanotechnology applications that can address societal challenges in energy, electronics, sensors and health.^{1,2,3} One closely related class of 2D crystals are the single-element “2D-Xenes”, which take their etymology from the sp^2 -hybridized alkene bond and are generally comprised of a single-layer of atoms organized into a honeycomb-like lattice. In particular, 2D-Xenes made of group-IVA elements, referred to as “silicene”, “germanene”, and “stanene” when $X=Si$, Ge , and Sn respectively, are isoelectronic to graphene,⁴ however their larger interatomic distances result in up and down atomic buckling about a honeycomb lattice⁵, and also afford new routes towards covalent functionalization. The buckled structure and the strong spin-orbit coupling (SOC) concurrently paves a pathway for accessing the new quantum state of matter known as a 2D topological insulator (TI), a solid-state material that has both an insulating electronic structure in the bulk and dissipation-less conducting channels along the edges that are protected against back-scattering by time reversal symmetry. The potential technological value of such an effect is that it enables the creation of a material that exhibits ideal conduction along the edges that is immune from scattering from a variety of defects and disorder. Overall, intriguing questions have come to the forefront such as: What are the benefits and challenges that arise owing to the intrinsic buckling and surface sensitivity in 2D-Xenes? And can 2D-Xenes ultimately establish topology as a paradigm shift for nanoscale opto-electronics? These questions are discussed here with highlights on the opportunities and perspectives on the technical challenges.

A pseudo planar flatland

While the planar flatland has been considered the norm for the past ten years as is the case for graphene and h-BN, *anisotropy* in honeycomb lattices stemming from the extended vertical distortions (for example bond buckling in 2D-Xenes or puckering in phosphorene) is now the *new normal*.^{4,5,6} The native buckling can have dramatic implications in the overall physics and chemistry of the 2D-Xenes. For instance, the degree of buckling can dictate the electronic character or influence the chemical reactivity. This highlights the concept of a pseudo-planar flatland where buckling can foster a number of potential functionalities that can be readily engineered.

An in-depth view on buckling. In their most stable form, 2D-Xenes based on Si, Ge or Sn atoms adopt a buckled hexagonal honeycomb structure^{6,7}, unlike graphene, which is ideally flat. For example, the larger bond length in silicene (~ 2.28 Å) compared to graphene (~ 1.42 Å), prevents the Si atoms from forming strong π -bonds, thus leading to deviations away from a sp^2 hybridization.^{4,8} The buckling of the Si atoms brings them closer together to enable a stronger overlap of their π -bonding p_z orbitals, resulting in a mixed sp^2 - sp^3 hybridization, which stabilizes their hexagonal arrangement. The vertical buckling distance δ between the top and bottom atoms of the 2D-Xene crystal structure (**Figure 1a**) is correlated to the bond angle between the framework atoms and the hybridization of the atomic orbitals: δ increases from 0 Å in graphene to 0.85 Å in free-standing (FS) stanene, with an associated decrease of the bond angle, from 120° (pure sp^2 hybridization of C atoms) to 110° (almost pure sp^3 hybridization of Sn atoms).

100 2D-Xenes can be described by a four-band second-nearest neighbor tight binding
 101 Hamiltonian⁹ near the K and K' points, involving a SOC term (λ_{SO}) and a staggered
 102 external electric field term (E_z),

$$103 \quad H = \hbar v_F (\eta k_x \tau_x + k_y \tau_y) + \eta \tau_z \sigma_z \lambda_{SO} - \frac{\delta}{2} E_z \tau_z + \frac{\lambda_{R1} (\eta \tau_x \sigma_y - \tau_y \sigma_x)}{2} + \eta \tau_z \lambda_{R2} (k_y \sigma_x -$$

$$104 \quad k_x \sigma_y), \quad (1)$$

105 where v_F is the Fermi velocity, $\eta=1$ for K and $\eta=-1$ for K' point, $\tau_{x,y,z}$ is the Pauli matrix
 106 for the sublattice pseudospin, σ is the Pauli matrix for the electron spin, and λ_{R1} and λ_{R2}
 107 are the first and second Rashba coupling constants, related to the nearest and second
 108 nearest neighbor hopping, respectively. E_z arises from the buckling of 2D-Xenes and
 109 allows for tunability of their electronic properties.

110 When the SOC is not taken into account and no electric field is applied (Kane-Mele
 111 model¹⁰), silicene, germanene, and stanene are predicted to be zero-gap semiconductors
 112 with linear dispersions near the K and K' Dirac points. The computed Fermi velocities in
 113 silicene and germanene are about 6×10^5 m/s,^{6,11} roughly half of that for graphene. With
 114 SOC, silicene and germanene are predicted to be 2D TIs with bulk energy gaps of about
 115 1.5-2 meV and 23.9-30 meV, respectively.⁹ Stanene, owing to its stronger SOC,
 116 incorporates a 2D TI state with a sizeable energy gap of about 0.1 eV.^{12,13}

117 **The 2D-Xene experimental Odyssey.** The tendency of FS 2D-Xenes to buckle is
 118 experimentally confirmed by the *epitaxial growth* of various 2D-Xene “phases” on
 119 different substrates with *commensurate* surface lattice. An extensive list of substrates
 120 serving this purpose is enumerated in **Table 1**. Experimental identification of the buckled
 121 2D-Xene structure has been mainly based on scanning tunneling microscopy (STM)

assisted by *ab initio* atomic models. Other techniques such as angle resolved photoemission spectroscopy (ARPES) have been used to reveal the electronic structure. Starting from the ideal FS 2D-Xene lattice as predicted from *ab initio* simulations (**Figure 1a**), a historical overview of the main experimental breakthroughs is presented in **Figure 1** including the STM confirmation of silicene (**b**), closely paralleled by the chemical synthesis of germanane (**c**), and followed by the epitaxy of germanene (**d**), the realization of a silicene transistor (**e**) and the epitaxy of stanene (**f**).

The forerunner of the 2D-Xene roadmap is the epitaxial silicene on metallic substrates.^{14,15-19} In particular, silicene-on-Ag(111) is interesting as a model of a buckled elemental atomic sheet due to the wealth of commensurate phases on the host Ag(111) surface.^{16,17} These phases are commonly written as $n \times n / m \times m$, where $n \times n$ and $m \times m$ refer to the coincidence with the FS 2D-Xene lattice and the substrate lattice, respectively; they arise from different rotational alignment of the 2D-Xene sheet with respect to the substrate. STM topographies of the majority silicene phases are reported in **Figure 1b** along with the respective atomic structure model. This structural versatility can dramatically influence the electronic structure of each 2D-Xene phase, and is a general monolayer property of the 2D-Xene family (**Table 1**).²⁰ In spite of this, substrate orbitals strongly hybridize with the Si atoms, resulting in an overall metallic character of the silicene-on-metal substrate system, as follows from ARPES investigations.^{11,21}

Following silicene, several STM studies have detailed the atomic arrangement of buckled germanene phases on various metallic substrates^{22,23,24-26} (**Figure 1d**), and the epitaxy of stanene on Bi₂Te₃ substrates has been established soon afterwards²⁷ (**Figure 1f**). Most recently, the epitaxy of a 2D boron sheet, termed borophene, with a metallic character

^{28,29} and of a monolayer of (silicene-like) phosphorene with semiconducting character³⁰ were respectively reported on Ag(111) and Au(111) substrates, thus extending the realization of 2D-Xenes to elements beyond the group IVA atoms.

Covalently functionalized 2D-Xenes. The covalent functionalization into ligand-terminated Xenes is an intriguing route towards tuning chemical, physical, and topological properties of these materials, as well as increasing their environmental stability. Here we define “Xanes” as 2D-Xene derivatives in which every atom on the framework features a covalently bound ligand, thereby removing π -bonding. These ligands directly couple to the half-filled p_z orbitals in 2D-Xenes, to produce a gapped, semiconducting band structure, and the magnitude of this band gap depends on both the framework element, and the identity of the covalent ligand.^{12,31,32} Unlike Xenes, Xanes can be derived by topotactic exfoliation.³¹ The first reported example of topotactic exfoliation of a single germanane sheet is shown in **Figure 1c**.

The most versatile route for preparing Xanes relies on the direct chemical conversion of structurally related layered intermetallic “Zintl” phase precursors. For example, the crystal structures of CaSi_2 , CaGe_2 , and BaSn_2 , are comprised of anionic $[\text{Si}]_n^-$, $[\text{Ge}]_n^-$, and $[\text{Sn}]_n^-$ layers in a puckered honeycomb arrangement analogous to 2D-Xenes, with each layer separated by a divalent cation. These phases can be transformed directly into multi-layered van der Waals crystals of Xanes by reaction with an electrophilic species that will bond directly to the group IVA framework and a solvent that solubilizes the divalent cation. For instance, the reaction of CaSi_2 in aqueous HCl produces partially –H/–OH terminated silicane, a reaction first reported by Friedrich Wöhler in 1863.³³ In

aqueous acidic conditions, the Si- and Sn-based Xane frameworks become partially terminated with -OH ligands, whereas the Ge-based frameworks are completely -H terminated.^{31,32} Organic-terminated Xanes can be created via a similar one-step reaction of precursor Zintl phases with alkyl halides. For instance, GeCH_3 has been prepared via the direct reaction of CaGe_2 with CH_3I .³⁴ Also, some Si and Ge surface functionalization routes can also graft ligands onto 2D Xanes, such as the hydrosilylation of alkenes and alkynes onto 2D silicane.³⁵ Other strategies for preparing Xanes include electrochemical atomic layer epitaxy, which has been shown to be a viable route towards germanane,³⁶ as well as the H-chemisorption to 2D-Xenes.³⁷

The handful of emerging reports on the role of surface termination on electronic properties of functionalized Xenes indicate that both the size and the electron-donating/electron-withdrawing properties of the ligand have a strong influence on the electronic structure. First, surface-terminating Xenes with ligands that have relatively larger van der Waals radii can produce tensile strain in the 2D framework. This tensile strain will lower the energy of the ligand-Si/Ge/Sn s-orbital antibonding levels at the conduction band maximum (CBM), effectively reducing the band gap.¹² Increasing the alkyl length by just a single methylene unit to ethyl ($\text{-CH}_2\text{CH}_3$) limits the fraction of surface atoms that are functionalized with the organic substituent to functionalization to 80-85% whereas the remaining atoms are -H terminated.³⁸ While the framework of functionalized Xenes might have the flexibility to accommodate the strain induced by larger ligands, to date, GeCH_3 is the only synthesized Xene functionalization with nearly uniform organic functionalization.³⁴ Second, more electronegative ligands will withdraw electron density from the Xane framework, and similarly lower the energy of the ligand-

Si/Ge/Sn s-orbital antibonding levels at the CBM.³⁹ Consequently, replacing the –H ligand in germanane with the more electron-donating –CH₃ ligand will increase the observed band gap from 1.54 to 1.67.^{12,34} These Xenes can be in the form of multilayered crystals whose length and width is determined by the size of precursor Zintl phases and can be greater than millimeters, suitable for exfoliation and further device processing.³¹

The triple S-factor in the epitaxial 2D-Xenes. The experimental realization of the epitaxial Xenes has been generally conditioned by three factors: Synthesis, Substrate, and Stability. This *triple S-factor* brings several challenges for a full exploitation of Xene and access to a broad range of characterization and/or applications. The synthesis of 2D-Xene is restricted to expensive molecular beam epitaxy (MBE) methodologies requiring ultrahigh vacuum conditions. Moreover, substrates enabling the 2D-Xene epitaxy must be commensurate with the FS 2D-Xene structure.^{14,15,16,17,19} Further efforts are in progress to design or synthesize 2D-Xenes on weakly interacting substrates, and the option of chalcogenide templates with a selected termination is promising to host a 2D-Xene lattice.⁴⁰ This is the case of the growth of hexagonal silicene sheets on semiconducting MoS₂.^{41,42} The option of Xenes on non-metallic substrates would bring obvious benefits for device integration, scientific studies, and applications, and *substrate engineering* is a demanding task of technological interest.

On the other hand, the chemical stability of 2D-Xenes under ambient condition is severely restricted by the inherent thermodynamic preference of Si, Ge, and Sn for *sp*³ bonding. Air-stability issues can be overcome by making use of a non-destructive encapsulation material such as Al₂O₃.⁴³

The case of silicene can be taken as a paradigmatic pathway to overcome the “triple S factor” and develop a prototypical 2D-Xene nanoelectronic platform (**Figure 1e**). A silicene sheet can be sandwiched in between two thin layers, the native Ag on one side and Al₂O₃ encapsulation on the other side. The former layer comes from the delamination of the “substrate”, a cleavable Ag(111)-on-mica where the silicene is epitaxially grown. The latter one is the stabilizing encapsulation sequentially deposited after silicene growth. The “silicene sandwich” is easily handled as a sheet to be placed upside down on a device-friendly platform, such as SiO₂/Si, and the top Ag patterned so as to obtain a Ag-free silicene channel in between two (or more) “native” Ag electrodes, hence completing a transistor device structure. Room temperature operation of this silicene transistor revealed an ambipolar character in agreement with the predicted graphene-like nature of the particular silicene phase in the experiment.⁴⁴ Despite the fast degradation of exposed Ag-free silicene in ambient condition that is yet to be fully addressed, this transfer methodology can be generally extended to the class of Xenes-on-substrate, and hence, offer a universal route for verifying the quantum transport properties of Xenes.⁴⁵

Topology as a paradigm shift for quantum electronics

2D TIs support the intriguing quantum spin Hall (QSH) effect, characterized by insulating bulk states with gapless helical edge states that are protected against back-scattering by time reversal symmetry (**Figure 2a**). The QSH effect was first theoretically predicted and subsequently experimentally observed in a HgTe quantum well.^{46,47} Graphene was proposed to have the same effect,¹⁰ but its bulk gap is extremely small (~μeV). The buckled 2D-Xenes have much larger QSH gaps (up to 0.3 eV),¹² suitable for

room-temperature usage, and their QSH states are tunable by chemical functionalization from π orbital states to σ orbital states (**Figure 2b**).¹² This tunable topology affords an electronic switch where the logic state is dictated by the topological phase transition rather than by the charge depletion/inversion as in conventional semiconductors.

The benefit of gaining weight. 2D-Xenes share the same Dirac physics as graphene, which is a 2D Dirac semimetal when excluding the SOC and becomes a QSH insulator when including the SOC (**Figure 2b**, left).^{10,9,12} Notably, the SOC-induced bulk gap in silicene (1.5-2 meV)^{45,9} is much larger than in graphene ($\sim\mu\text{eV}$). The underlying mechanism is two-fold. The first-order contribution of SOC, which is turned off in a planar structure like graphene, is turned on in silicene because the hybridization between π and σ orbitals of nearest neighbors becomes allowed in a buckled structure.⁹ Moreover, the SOC strength is commensurate with increasing weight. Thus enhanced SOC is obtained in germanene (23.9-30.0 meV) and stanene ($\sim 0.1\text{eV}$).^{9,12,48} In perspective, the QSH gap increases five orders of magnitude from graphene to stanene, compliments of the heavier mass and structural buckling.

The chemical functionalization of 2D-Xenes provides an additional degree of freedom to tune topological states. The saturation of the p_z orbitals opens a large energy gap at the K and K' points, which destroys the Dirac physics of the π orbitals. The low-energy physics is then determined by the σ orbitals around the Γ point (**Figure 2b**, right), where inverted band structures may occur in 2D-Xenes composed of heavier elements.^{12,48} Take the fluorinated stanene (F-stanene) as an example. The relevant states around the Fermi level are comprised of s , p_x , and p_y orbitals on the Sn atoms, split into bonding and anti-bonding states by the Sn-Sn interaction. Unlike undecorated graphene, the anti-bonding s

state is located below the bonding p_{xy} state at the Γ point, showing an inverted band order. The s - p band inversion results in a nontrivial QSH phase,¹² similar to the scenario of the HgTe quantum well.⁴⁶ Therefore, the QSH states are varied from π orbitals described by the Kane-Mele model¹⁰ to σ orbitals described by the Bernevig-Hughes-Zhang model⁴⁶. The new QSH states derived from σ orbitals possess many advantages^{12,48}. First, the QSH gaps are extraordinarily large (up to 0.3 eV), since the effective SOC of σ orbitals is significantly larger than that of π orbitals. Second, the electronic properties can be controllably tuned. Functionalization with different chemical groups can result in topologically distinct phases (**Figure 2c**). Moreover, applying external strain can effectively control the emergence or disappearance of the helical edge states. Third, in comparison with the π orbitals that are easily affected by substrate and adsorbates,^{27,49} the σ orbitals that are composed of in-plane p_{xy} states are relatively more environmentally stable. Even when forming chemical bonds with the substrate, large-gap σ -orbital-derived QSH states are predicted to be preserved in functionalized stanane⁴⁰. Given the dependence on external factors owing to their buckled nature, 2D-Xenes may offer a high degree of functional versatility. For example, in silicene, physical influences such as electric and magnetic fields, mechanical strain, and temperature has been predicted to produce a variety of electronic states including semimetallic, semiconducting, superconducting, and trivial insulating phases, in addition to about *sixteen* 2D TI phases.⁵⁰ These TI phases arise from the combination of the four defining topological numbers in a honeycomb lattice, which are the Chern, spin-Chern, valley-Chern and spin-valley-Chern numbers with an end result of five types of topological quantum states, namely the spin-polarized quantum anomalous Hall (SQAH), quantum

valley Hall (QVH) and quantum spin-valley Hall (QSVH) states in addition to the already mentioned QSH and QAH states (**Figure 2d**).⁵¹

A Buckled Outlook

The interplay of non-trivial 2D topological features and the buckled structure of 2D-Xenes may also open new routes for disruptive applications like low-energy electronics, piezo-magnetism, enhanced thermoelectricity and spintronics.¹² Furthermore, by explicitly breaking the symmetries (for example, gauge symmetry, inversion symmetry and time reversal symmetry), new emerging physics (such as topological superconductivity⁵², valley-polarized metal,⁴⁵ and QAH effect⁴⁵) can be explored in 2D-Xenes.

From digital to topological bits. One of the more visionary outlook about 2D-Xenes is the realization of a paradigm shift from conventional to quantum electronics based on field-controllable topological properties. Transport through the QSH edge states can be ballistic, thus promising an exceptional advance in terms of energy saving in electronic devices. In contrast to carbon nanotubes or graphene where ballistic transport can be affected by defects, transport in 2D-Xenes in the QSH regime is robust against scattering. Equally appealing is the opportunity to access a logic functionality based on the concept of topological phase transitions. A topological phase transition occurs when the QSH state characterized by ballistic edge states that can serve as information carriers (“ON” state) is switched to a trivial insulator state with vanishing conductance (“OFF” state) vis-à-vis an external electric field E_z (**Figure 3a**, top diagram).⁵⁰ The two ON/OFF states define a logic switch mechanism, namely a *topological bit*, which can be distinguished by

the sequential topological gap closing and trivial gap opening as a function of the external field.

A prototypical device structure based on the gate bias modulation can be understood as a *quantum* topological insulator field-effect transistor (TI-FET)⁵³ as shown in the corresponding sketch of **Figure 3b**. The gate bias can be employed to switch the TI-FET from the ON to the OFF state, whereas the drain-source voltage facilitates one kind of helical carrier through each edge. Being topologically protected and helical in character, the edge states in these quantum TI-FETs serve as truly one-dimensional, spin-polarized dissipationless highways for charge transport that is robust against scattering from non-magnetic material defects,¹² and potentially suitable for scaled low-power nanoelectronic devices.

The main challenge with devices based on electric field-driven topological phase transitions is the relatively large critical field required ($E_c = 2\lambda_{SO}/\delta$, where $2\lambda_{SO}$ is the SOC induced bandgap), which is of the order of 0.05 V/nm and 1 V/nm for silicene and stanene, respectively. In light of this, a *classical* TI-FET not predicated on topological phase transitions was recently proposed.⁵⁴ In this scheme, the 2D TI is configured as a nanoribbon FET with the width small enough to ensure inter-edge elastic scattering (namely, within an order of magnitude of the field-dependent edge-bulk tunneling length scale, ~ 1 nm for F-stanene and ten times larger in silicene). Unlike the conventional FET where the field-effect enhances the carrier density and conductivity increases, the field-effect in classical TI-FET facilitates phonon-mediated scattering of edge electrons into bulk states by reducing the energy barrier and hence, conductivity decreases.

Alternatively, a topological phase transition can be driven by applying external strain in functionalized 2D-Xenes because the *s-p* band inversion at the Γ point sensitively depends on the interatomic bond length.¹² In this case, the details of the topological phase transition depend on the identity of the functional group. For instance, F-stanene is a QSH insulator that undergoes a transition to a trivial insulator state as a function of the compressive strain (over a critical value of about -7%) thus nominally mimicking the same ON/OFF sequence of the electric field induced topological phase transition (**Figure 3a**, middle diagram). Conversely, H-stanene is a trivial insulator at zero applied strain and enters a QSH state when applying a tensile strain over a critical value of about 2% thus manifesting an inverted OFF-ON logic mechanism (**Figure 3a**, bottom diagram). Both mechanisms may open unprecedented routes for 2D-Xene-based piezotronic devices (see the corresponding sketches in **Figure 3b**).

New emerging applications.

The existence of large-gap, tunable and robust QSH states make 2D-Xenes promising for various applications (**Figure 3c**). The large bulk gap (~ 0.3 eV)¹² enables QSH phenomena to occur at room temperature. The helical edge states are useful for low-energy electronics, since they can conduct electricity without generating heat^{10,46,47}. They can also induce anomalous Seebeck effect due to the strong energy dependence in lifetime introduced by the edge-bulk interaction.⁵⁵ The thermoelectric performance of QSH insulators can be effectively improved by optimizing the morphology to maximize the edge-state contribution.⁵⁵ In addition, the helical edge states may also find important applications in spintronics (for example in spin transfer torque, spin Seebeck, spin-charge conversion, and spin-filter⁵⁶ based devices). However, very little experimental progress has been achieved until now, mainly because high-

quality samples are still lacking.²⁷ More experimental effort is thus essential to advance 2D-Xenes and related materials, and ultimately realize the above-mentioned concepts.

New emerging physics. 2D-Xenes may also offer an ideal platform to study new emerging physics. Exotic quantum effects would emerge upon explicitly breaking different symmetries in 2D-Xenes (**Figure 3c**). For instance, breaking the *inversion symmetry* by substrate or electric field introduces the Rashba effect, which enables the spin manipulation as well as access to topological superconductivity.⁵² Breaking *time reversal symmetry* by magnetism^{40,57} may induce the QAH effect, which supports chiral edge states that are topologically protected against scattering. A near room-temperature QAH effect has been predicted in half-functionalized stanene and germanene, where a ferromagnetic order is formed between local magnetic moments of unsaturated p_z states.⁵⁷ Breaking *gauge symmetry* by superconductivity may afford exotic topological superconducting states bearing Majorana fermions that are useful for fault-tolerant topological quantum computation.⁵² Even richer physics is expected by breaking different symmetries simultaneously. For instance, the use of an electric field together with an exchange field ΔM (from magnetic impurities or by proximity coupling with a ferromagnet) enables tuning the spin and valley degrees of freedom independently, which generates a variety of phases in 2D-Xenes.⁴⁵ Furthermore, the symmetries can be broken either globally or locally. The local symmetry breaking, though preserving the topology, can considerably affect properties of the QSH states. As an example, the helical edge states are eliminated when introducing an in-plane ferromagnetic order to the edge atoms. The existence of the helical edge

states thus can be controlled by local edge manipulation, which leads to potential applications in giant magnetoresistance and an ideal spin filter.⁵⁶

Perspective

By virtue of their buckled character, 2D-Xenes can potentially serve as a nanotechnology platform with multiple physical features for fundamental research and applications. The diverse material properties in intrinsic and functionalized Xenes permit a wide degree of freedom in designing efficient, on-demand or reconfigurable devices for electronic, spintronic, photonic, thermal, energy, mechanical, chemical and sensor nanosystems. However, many processing and technological issues are yet to be solved, the most significant of which is the instability in ambient conditions which complicates the establishment of scalable routes for device integration schemes. Appropriate solutions including the design of device-friendly and neutral substrates, and the development of large-scale production methodologies (for example chemical vapor deposition) are demanding issues for a viable Xene nanotechnology. Nonetheless, the recently reported technological and synthetic advances are encouraging towards the exploration of the unique topological phenomena in 2D-Xenes and their exploitation in innovative device concepts.

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416 **Item Captions**

417 **Table 1. Properties of 2D-Xenes.** Computed structural parameters of FS 2D-Xenes,
418 using first-principles calculations based on density functional theory (left-hand side).
419 Experimentally observed or calculated structural parameters of epitaxial Xenes-on-
420 substrates (right-hand side). a_X , ℓ_{X-X} , and δ are the cell lattice constant, interatomic
421 distance, and buckling parameter in 2D-Xenes. E_g is the SOC-induced band gap. The
422 latter branch includes a list of substrates that have been found to accommodate specific
423 2D-Xene lattices and their atomic reconstructions (commonly denoted as “phases”) on
424 the substrates’ surface. Graphene parameters are reported for comparison. Note that
425 graphene has nominally zero atomic buckling in the absence of rippling.

426

427 **Figure 1: The 2D-Xene Odyssey.** An evolutionary overview of the key-
428 experimental evidences of 2D-Xenes. (a) Atomic configuration of a FS 2D-Xene lattice
429 (top: side view, bottom: perspective view) where red and orange spheres represent X
430 atoms at top and bottom positions, respectively. (b) Silicene epitaxy on Ag(111)
431 substrates as a representative case of Xene-on-substrate systems. STM investigations of
432 Silicene-on-Ag(111) has revealed a multiphase character. Left: The STM topographies of
433 three majority phases, $3\times 3/4\times 4$, $\sqrt{7}\times\sqrt{7}/\sqrt{13}\times\sqrt{13}$, and $\sqrt{7}\times\sqrt{7}/2\sqrt{3}\times 2\sqrt{3}$, are reported
434 along with the atomic structure models. (c) Topotactic exfoliation of germanane as a first
435 example of the chemical exfoliation of a Xane sheet: process schematics (left) and optical
436 microscopy image of a single hydrogen-terminated germanane sheet (right). (d)
437 Representative case of germanene on metal substrates: STM topography of an extended

germanene sheet on Al(111) assisted by an *ab initio* atomic model. Adapted from Ref.²⁴.
 (e) Silicene integration into a transistor device after Ref.⁴⁴: Silicene is epitaxially grown
 on a cleavable Ag-on-mica substrate via MBE, encapsulated with a protective Al₂O₃ and
 then delaminated as a Al₂O₃/silicene/Ag sandwich sheet; this “silicene sandwich” is
 transferred to a device-friendly substrate (such as SiO₂/Si) and then integrated as the
 active channel in a field-effect transistor structure by patterning the native Ag to make
 contact electrodes. (f) Epitaxy of stanene on Bi₂Te₃ substrates based on STM
 identification of a buckled and unreconstructed lattice structure. Adapted from Ref.²⁷.

Figure 2: Topology as a paradigm shift in nanoelectronics. **a)** Top view of a 2D-
 Xene in a QSH state bearing helical currents at the edges. **b)** Side view of the atomic
 structure and molecular orbital, and the low-energy bands with (red solid lines) and
 without (black dashed lines) spin-orbit coupling for pure (left) and functionalized (right)
 Xene. The red and blue spheres denote atoms of different sublattices, and the white
 spheres denote the chemical functional group. **c)** Energy gap as a function of the lattice
 constant induced by the covalent functionalization of stanene. Adapted from Ref.¹². **d)**
 The topological phase diagram as a function of perpendicular electric field E_z and the
 exchange field ΔM , and the arrows reflect the spin polarization. Adapted from Ref.⁵¹

Figure 3: The topological bit and emerging physics based on broken symmetry.
a) Schematic band structure diagrams of nanoribbons along the 1D Brillouin zone (Γ -X',
 X' is a k -point point along the nanoribbon length direction) for (top) bare stanene,
 (middle) and fluorine-terminated (SnF), and (bottom) hydrogen-terminated (SnH) stanane

and topological phase transitions induced by an out-of-plane electric field or a compressive/tensile strain. The helical edge states that emerge within the bulk gap are denoted by red lines. **b)** Corresponding sketches of a *quantum* TI-FET driven by an electric field, by a compressive strain, and by a tensile strain. A 2D-Xene ribbon is embedded in between a top gate and a substrate (for applying V_G), and source (S) and drain (D) electrodes. **c)** Applications of the QSH states and new emerging physics induced by breaking symmetries, including the A-B sublattice symmetry, inversion symmetry P , time reversal symmetry T and gauge symmetry U .

FS 2D-Xenes (predicted)						Epitaxial 2D-Xenes (Xene-on-substrate)					
	a_X (Å)	ℓ_{X-X} (Å)	δ (Å)	E_g	Ref.	substrate	phase	a_X (Å)	ℓ_{X-X} (Å)	δ (Å)	Ref.
Graphene	2.47	1.42	0	~1 μeV		metals		2.47	1.42	0	58
Silicene	3.87	2.28	0.44	1.5-2 meV	59	Ag(111)	3×3/4×4	11.78	2.34-2.39	0.71- 0.79	14,15 , 16,17
							√7×√7/√13×√13	10.60	2.31-2.36	0.77- 0.79	
							√7×√7/2√3×2√3	10.20	2.28-2.37	1.10	
						Ir(111)	√3×√3/√7×√7	7.20	2.10	0.83	18
						ZrB ₂	√3×√3/2×2	6.72- 6.78	2.24-2.26	0.90- 1.35	19
Germanene	4.06	2.44	0.69	23.9- 30.0 meV	59	MoS ₂	1×1/1×1	3.17	2.61	1.90	41
						Au(111)	√3×√3/√7×√7	7.65	2.55	0.47	22
						Pt(111)	3×3/√19×√19	12.0	~2.31	0.60	23
						Al(111)	2×2/3×3	8.50	2.58	1.23	24
						MoS ₂	1×1/1×1	3.82	2.20	0.86	42
Stanene	4.68	2.83	0.85	0.1 eV	12	hex-AlN	3×3/4×4	12.31	2.37	0.70	26
						Bi ₂ Te ₃	1×1/1×1	4.38	2.80	1.20	27

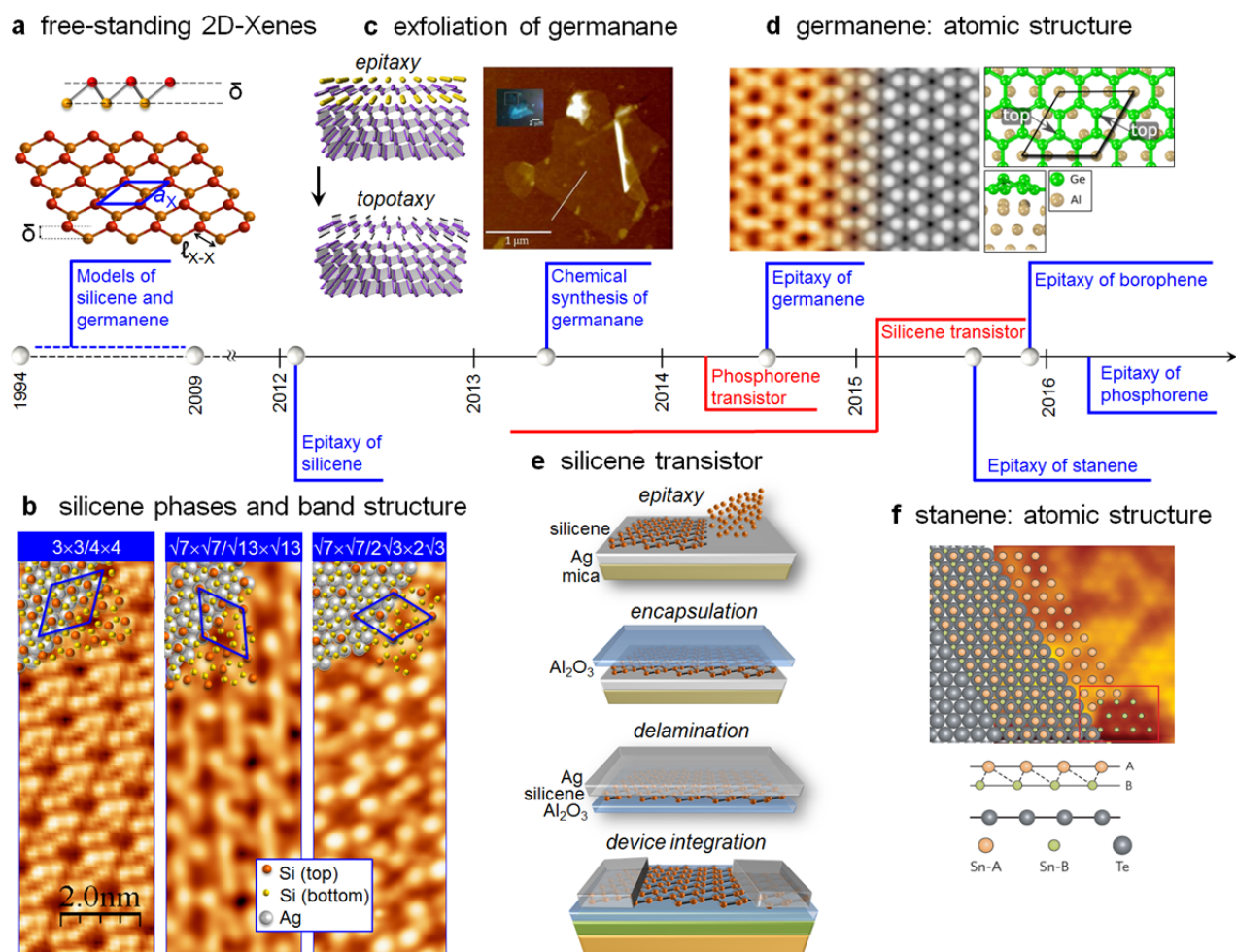


Figure 1

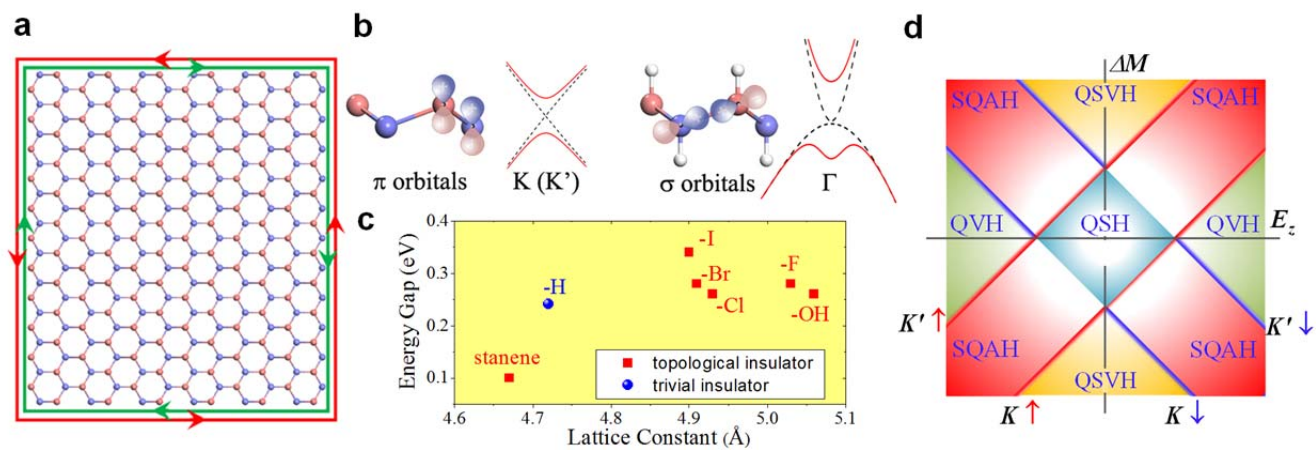


Figure 2

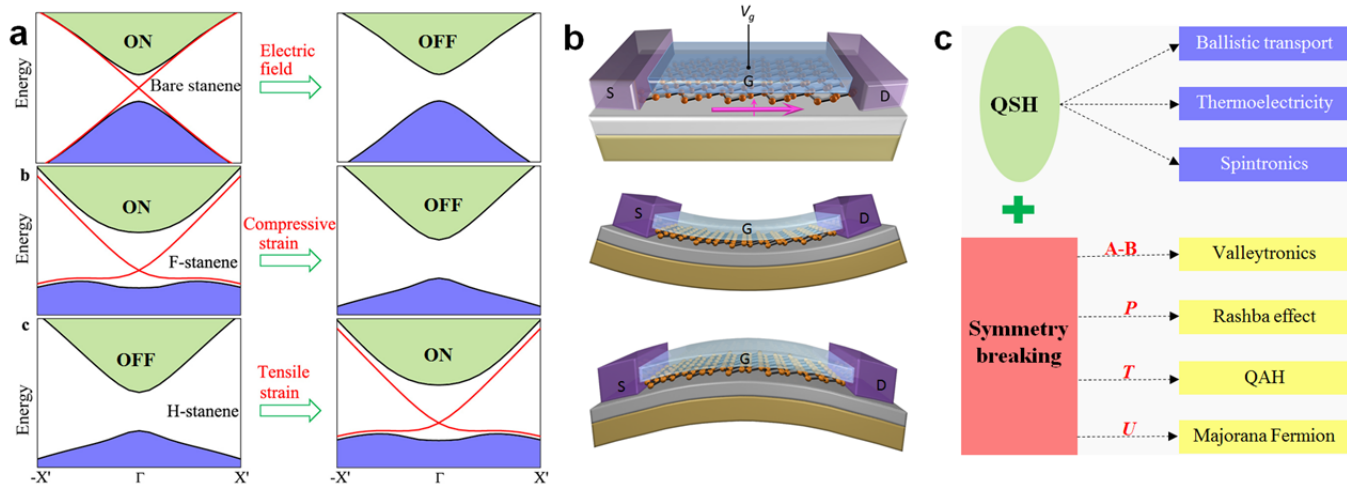


Figure 3

524 **List of References**

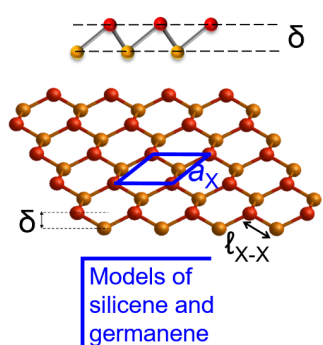
- 525 ¹ G Fiori, F Bonaccorso, G Iannaccone, T Palacios, D Neumaier, A Seabaugh, SK
526 Banerjee, and L Colombo, *Nature Nanotechnology* **9** (10), 768 (2014).
- 527 ² AC Ferrari, F Bonaccorso, V Fal'ko, KS Novoselov, S Roche, P Boggild, S
528 Borini, FHL Koppens, V Palermo, N Pugno, JA Garrido, R Sordan, A Bianco, L
529 Ballerini, M Prato, E Lidorikis, J Kivioja, C Marinelli, T Ryhanen, A Morpurgo,
530 JN Coleman, V Nicolosi, L Colombo, A Fert, M Garcia-Hernandez, A Bachtold,
531 GF Schneider, F Guinea, C Dekker, M Barbone, ZP Sun, C Galiotis, AN
532 Grigorenko, G Konstantatos, A Kis, M Katsnelson, L Vandersypen, A Loiseau, V
533 Morandi, D Neumaier, E Treossi, V Pellegrini, M Polini, A Tredicucci, GM
534 Williams, BH Hong, JH Ahn, JM Kim, H Zirath, BJ van Wees, H van der Zant, L
535 Occhipinti, A Di Matteo, IA Kinloch, T Seyller, E Quesnel, XL Feng, K Teo, N
536 Rupesinghe, P Hakonen, SRT Neil, Q Tannock, T Lofwandra, and J Kinaret,
537 *Nanoscale* **7** (11), 4598 (2015).
- 538 ³ Akinwande D, Tao L, Yu Q, Lou X, Peng P, and Kuzum D, *IEEE*
539 *Nanotechnology Magazine* **9**, 6 (2015).
- 540 ⁴ S Balendhran, S Walia, H Nili, S Sriram, and M Bhaskaran, *Small* **11**, 640
541 (2015).
- 542 ⁵ K Takaeda and K Shiraishi, *Physical Review B* **50**, 14916 (1994).
- 543 ⁶ S Cahangirov, M Topsakal, E Akturk, H Sahin, and S Ciraci, *Physical Review*
544 *Letters* **102**, 236804 (2009).
- 545 ⁷ GG Guzman-Verri and LCLY Voon, *Physical Review B* **76**, 075131 (2007).
- 546 ⁸ Carlo Grazianetti, Eugenio Cinquanta, and Alessandro Molle, *2d Materials* **3** (1)
547 (2016).
- 548 ⁹ CC Liu, WX Feng, and YG Yao, *Physical Review Letters* **107** (7), 076802
549 (2011).
- 550 ¹⁰ CL Kane and EJ Mele, *Physical Review Letters* **95**, 146802 (2005).
- 551 ¹¹ M Houssa, A Dimoulas, and A Molle, *Journal of Physics-Condensed Matter* **27**,
552 253002 (2015).
- 553 ¹² Y Xu, BH Yan, HJ Zhang, J Wang, G Xu, PZ Tang, WH Duan, and SC Zhang,
554 *Physical Review Letters* **111**, 136804 (2013).
- 555 ¹³ B van den Broek, M Houssa, E Scalise, G Pourtois, VV Afanas'ev, and A
556 Stesmans, *2D Materials* **1**, 021004 (2014).
- 557 ¹⁴ P Vogt, P De Padova, C Quaresima, J Avila, E Frantzeskakis, MC Asensio, A
558 Resta, B Ealet, and G Le Lay, *Physical Review Letters* **108**, 155501 (2012).
- 559 ¹⁵ CL Lin, R Arafune, K Kawahara, N Tsukahara, E Minamitani, Y Kim, N Takagi,
560 and M Kawai, *Applied Physics Express* **5**, 045802 (2012).
- 561 ¹⁶ BJ Feng, ZJ Ding, S Meng, YG Yao, XY He, P Cheng, L Chen, and KH Wu,
562 *Nano Letters* **12**, 3507 (2012).
- 563 ¹⁷ Daniele Chiappe, Carlo Grazianetti, Grazia Tallarida, Marco Fanciulli, and
564 Alessandro Molle, *Advanced Materials* **24**, 5088 (2012).
- 565 ¹⁸ L Meng, YL Wang, LZ Zhang, SX Du, RT Wu, LF Li, Y Zhang, G Li, HT Zhou,
566 WA Hofer, and HJ Gao, *Nano Letters* **13**, 685 (2013).
- 567 ¹⁹ A Fleurence, R Friedlein, T Ozaki, H Kawai, Y Wang, and Y Yamada-Takamura,
568 *Physical Review Letters* **108**, 245501 (2012).

569 20 E Scalise, E Cinquanta, M Houssa, B van den Broek, D Chiappe, C Grazianetti, G
 570 Pourtois, B Ealet, A Molle, M Fanciulli, VV Afanas'ev, and A Stesmans, *Applied*
 571 *Surface Science* **291**, 113 (2014).
 572 21 Tsoutsou D, Xenogiannopoulou E, Golias E, Tsipas P, and Dimoulas A, *Applied*
 573 *Physics Letters* **103**, 231604 (2013).
 574 22 ME Davila, L Xian, S Cahangirov, A Rubio, and G Le Lay, *New Journal of*
 575 *Physics* **16**, 095002 (2014).
 576 23 LF Li, SZ Lu, JB Pan, ZH Qin, YQ Wang, YL Wang, GY Cao, SX Du, and HJ
 577 Gao, *Advanced Materials* **26**, 4820 (2014).
 578 24 M Derivaz, D Dentel, R Stephan, MC Hanf, A Mehdaoui, P Sonnet, and C Pirri,
 579 *Nano Letters* **15**, 2510 (2015).
 580 25 P Bampoulis, L Zhang, A Safaei, R van Gastel, B Poelsema, and HJW Zandvliet,
 581 *Journal of Physics-Condensed Matter* **26**, 442001 (2014).
 582 26 F D'Acapito, S Torrenco, E Xenogiannopoulou, P Tsipas, J Marquez Velasco, D
 583 Tsoutsou, and A Dimoulas, *Journal of Physics - Condensed Matter* (2016).
 584 27 FF Zhu, WJ Chen, Y Xu, CL Gao, DD Guan, CH Liu, D Qian, SC Zhang, and JF
 585 Jia, *Nature Materials* **14**, 1020 (2015).
 586 28 AJ Mannix, XF Zhou, B Kiraly, JD Wood, D Alducin, BD Myers, XL Liu, BL
 587 Fisher, U Santiago, JR Guest, MJ Yacaman, A Ponce, AR Oganov, MC Hersam,
 588 and NP Guisinger, *Science* **350** (6267), 1513 (2015).
 589 29 Baojie Feng, Jin Zhang, Qing Zhong, Wenbin Li, Shuai Li, Hui Li, Peng Cheng,
 590 Sheng Meng, Lan Chen, and Kehui Wu, *Nat Chem* **advance online publication**
 591 (2016).
 592 30 Jia Lin Zhang, Songtao Zhao, Cheng Han, Zhunzhun Wang, Shu Zhong, Shuo
 593 Sun, Rui Guo, Xiong Zhou, Cheng Ding Gu, Kai Di Yuan, Zhenyu Li, and Wei
 594 Chen, *Nano Letters* (2016).
 595 31 Elisabeth Bianco, Sheneve Butler, Shishi Jiang, Oscar D. Restrepo, Wolfgang
 596 Windl, and Joshua E. Goldberger, *Acs Nano* **7**, 4414 (2013).
 597 32 Shishi Jiang, Maxx Q. Arguilla, Nicholas D. Cultrara, and Joshua E. Goldberger,
 598 *Accounts of Chemical Research* **48** (1), 144 (2015).
 599 33 F. Wohler, *Justus Liebigs Annalen der Chemie* **127** (3), 257 (1863).
 600 34 Shishi Jiang, Sheneve Butler, Elisabeth Bianco, Oscar D. Restrepo, Wolfgang
 601 Windl, and Joshua E. Goldberger, *Nature Communications* **5**, 3389 (2014).
 602 35 Hideyuki Nakano, Mitsuru Nakano, Koji Nakanishi, Daiki Tanaka, Yusuke
 603 Sugiyama, Takashi Ikuno, Hirotaka Okamoto, and Toshiaki Ohta, *Journal of the*
 604 *American Chemical Society* **134** (12), 5452 (2012).
 605 36 XH Liang, QH Zhang, MD Lay, and JL Stickney, *Journal of the American*
 606 *Chemical Society* **133** (21), 8199 (2011).
 607 37 Qiu J, Fu X, Xu Y, Zhou Q, Meng S, Li H, Chen L, and Wu K, (*ACS Nano*,
 608 2015), Vol. 9, pp. 11192.
 609 38 Hongbin Yu, Lauren J. Webb, Santiago D. Solares, Peigen Cao, William A.
 610 Goddard, III, James R. Heath, and Nathan S. Lewis, *Journal of Physical*
 611 *Chemistry B* **110**, 23898 (2006).
 612 39 CG van de Walle and JE Northrup, *Physical Review Letters* **70**, 1116 (1993).
 613 40 Y Xu, PZ Tang, and SC Zhang, *Physical Review B* **92**, 081112 (2015).

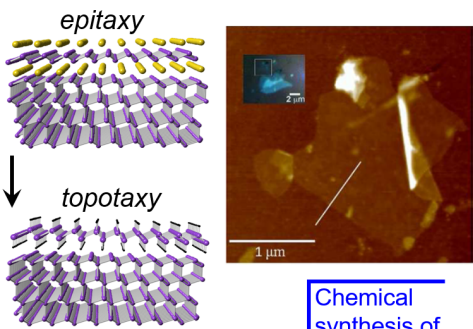
614 41 D Chiappe, E Scalise, E Cinquanta, C Grazianetti, B van den Broek, M Fanciulli,
 615 M Houssa, and A Molle, *Advanced Materials* **26** (13), 2096 (2014).
 616 42 L. Zhang, P. Bampoulis, A. N Rudenko, Q. Yao, A. van Houselt, B. Poelsema,
 617 M. I Katsnelson, and H. J W Zandvliet, *Physical Review Letters* **116** (25),
 618 256804 (2016).
 619 43 A Molle, C Grazianetti, D Chiappe, E Cinquanta, E Cianci, G Tallarida, and M
 620 Fanciulli, *Advanced Functional Materials* **23** (35), 4340 (2013).
 621 44 L Tao, E Cinquanta, D Chiappe, C Grazianetti, M Fanciulli, M Dubey, A Molle,
 622 and D Akinwande, *Nature Nanotechnology* **10** (3), 227 (2015).
 623 45 M Ezawa, *Physical Review Letters* **109**, 055502 (2012).
 624 46 BA Bernevig, TL Hughes, and SC Zhang, *Science* **314**, 1757 (2006).
 625 47 M Konig, S Wiedmann, C Brune, A Roth, H Buhmann, LW Molenkamp, XL Qi,
 626 and SC Zhang, *Science* **318**, 766 (2007).
 627 48 C Si, JW Liu, Y Xu, J Wu, BL Gu, and WH Duan, *Physical Review B* **89**,
 628 115429 (2014).
 629 49 PZ Tang, PC Chen, WD Cao, HQ Huang, S Cahangirov, LD Xian, Y Xu, SC
 630 Zhang, WH Duan, and A Rubio, *Physical Review B* **90** (12) (2014).
 631 50 Motohiko Ezawa, *Journal of the Physical Society of Japan* **84** (12), 121003
 632 (2015).
 633 51 M Ezawa, *Physical Review B* **87**, 155415 (2013).
 634 52 J Wang, Y Xu, and SC Zhang, *Physical Review B* **90**, 054503 (2014).
 635 53 M Ezawa, *Applied Physics Letters* **102**, 172103 (2013).
 636 54 WG Vandenberghe and MV Fischetti, *Journal of Applied Physics* **116**, 173707
 637 (2014).
 638 55 Y Xu, ZX Gan, and SC Zhang, *Physical Review Letters* **112**, 226801 (2014).
 639 56 S Rachel and M Ezawa, *Physical Review B* **89**, 195303 (2014).
 640 57 SC Wu, GC Shan, and BH Yan, *Physical Review Letters* **113**, 256401 (2014).
 641 58 J Wintterlin and ML Bocquet, *Surface Science* **603** (10-12), 1841 (2009).
 642 59 E Scalise, M Houssa, G Pourtois, B van den Broek, V Afanas'ev, and A Stesmans,
 643 *Nano Research* **6** (1), 19 (2013).
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							$\sqrt{7}\times\sqrt{7}/\sqrt{13}\times\sqrt{13}$	10.60	2.31- 2.36	0.77- 0.79	
							$\sqrt{7}\times\sqrt{7}/2\sqrt{3}\times 2\sqrt{3}$	10.20	2.28- 2.37	1.10	
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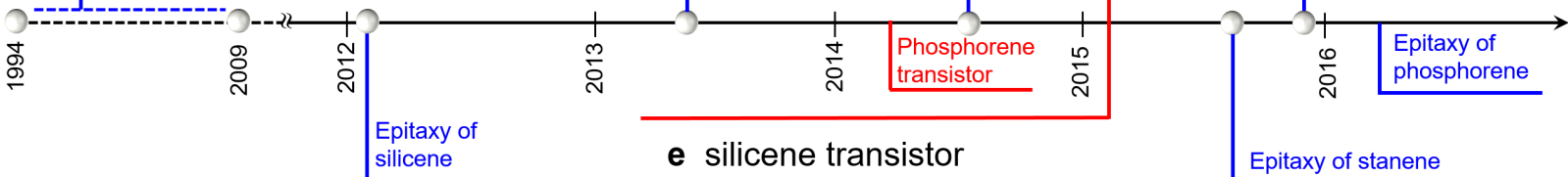
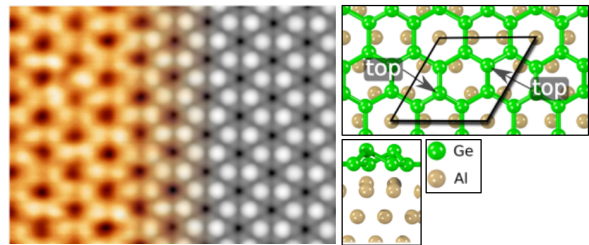
a free-standing 2D-Xenes



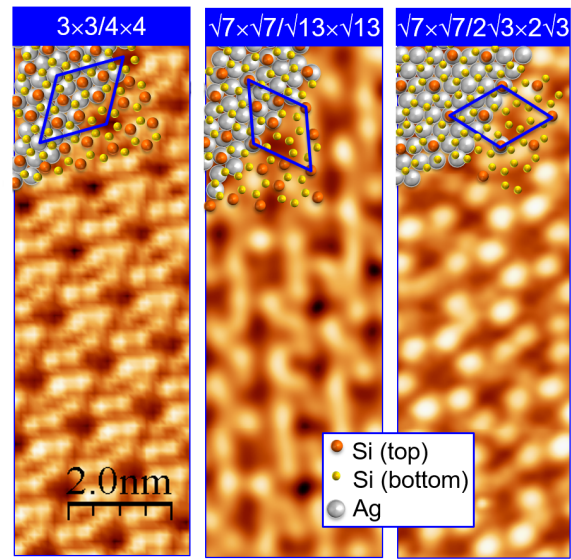
c exfoliation of germanane



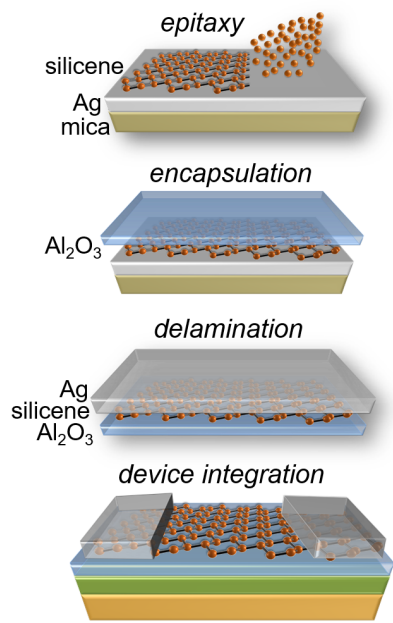
d germanene: atomic structure



b silicene phases and band structure



e silicene transistor



f stanene: atomic structure

